

# MnO<sub>2</sub> Resin

09/05/2008 - Madrid

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# Outline

- § pH dependency of  $D_w$
  - § Kinetics
  - § Capacity / Amount of resin
  - § Interferences
  - § Flow-rate
  - § Applications
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# Information on MnO<sub>2</sub>

Bill Burnett et al:

Use of de-ionised , ground and sea waters

§ Study with variation of

- pH
- Kinetics
- Ratio resin mass vs volume of solution
- Salinity
- Flow rates

§ Use of <sup>133</sup>Ba as Ra homologue

Josue Moreno (TU Munich):

§ D<sub>w</sub> values of natural radionuclides

§ Capacities

§ Interferences

# pH dependency

## Experimental

- § Batch experiments
- § 5 mL aqueous phase and 100 mg MnO<sub>2</sub> Resin OR  
10 mL aqueous phase and 25 mg MnO<sub>2</sub> Resin
- § pH adjustment with HCl or NaOH
- § Magnetic stirring, 60 minutes, room temperature
- § Phase separation by filtration (0,1 µm pores)

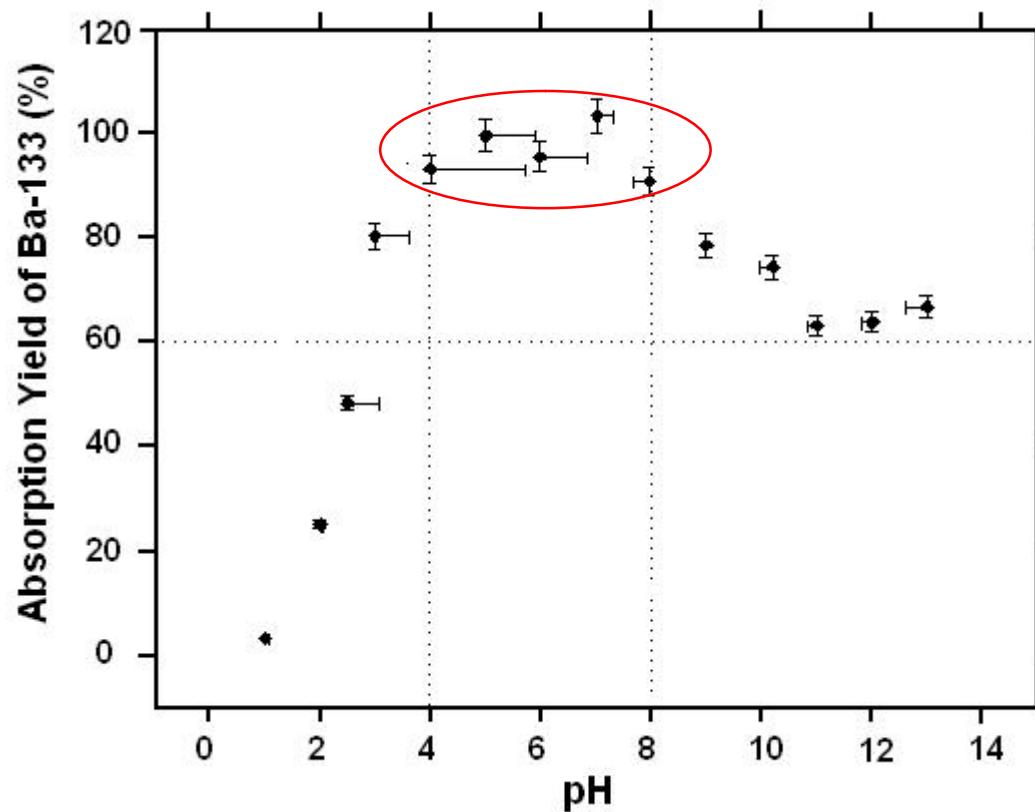
## Observation

- § Strong shift of the pH in aqueous phase

## Measurement

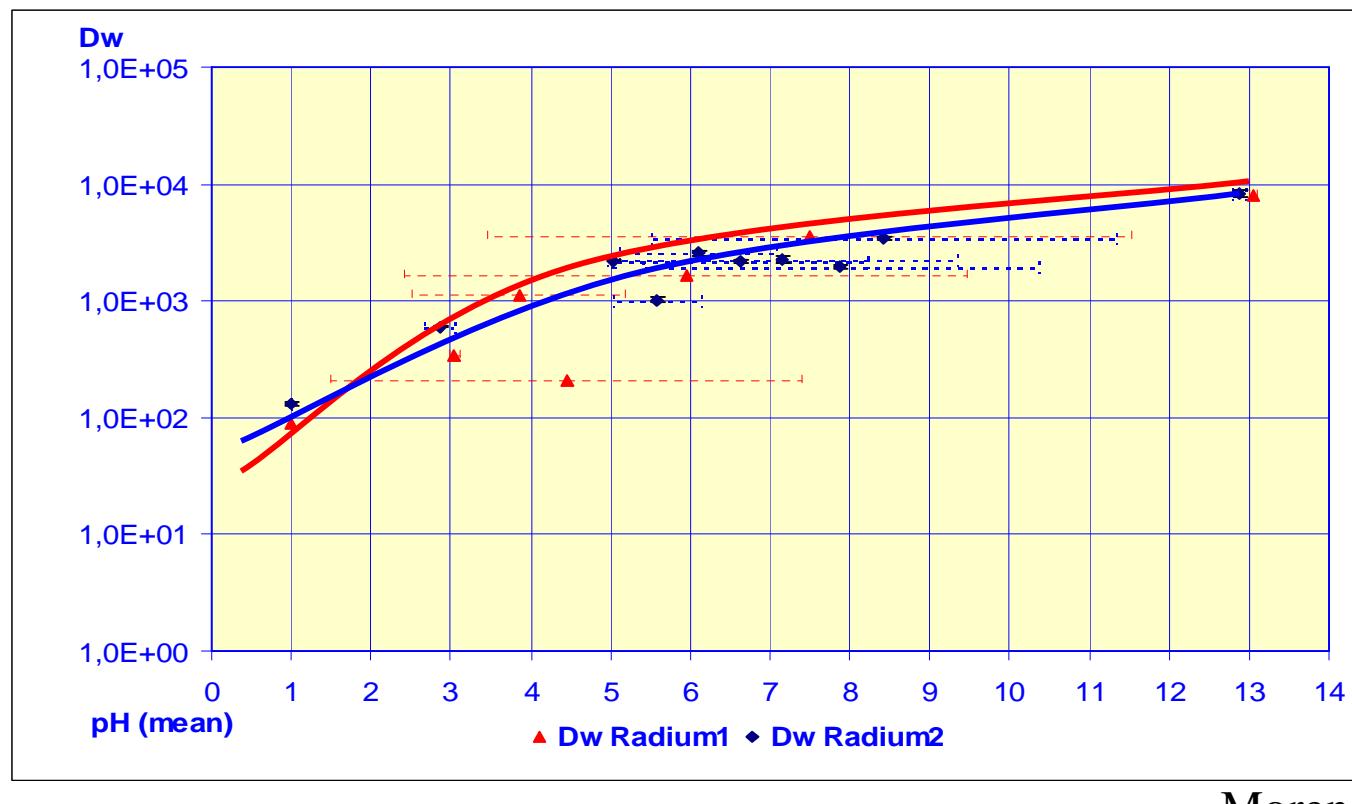
- § filtered MnO<sub>2</sub> via γ-spec or LSC (after resin elution)
- § aqueous phase via γ-spec or LSC

# pH dependency Ba-133



Burnett et al.

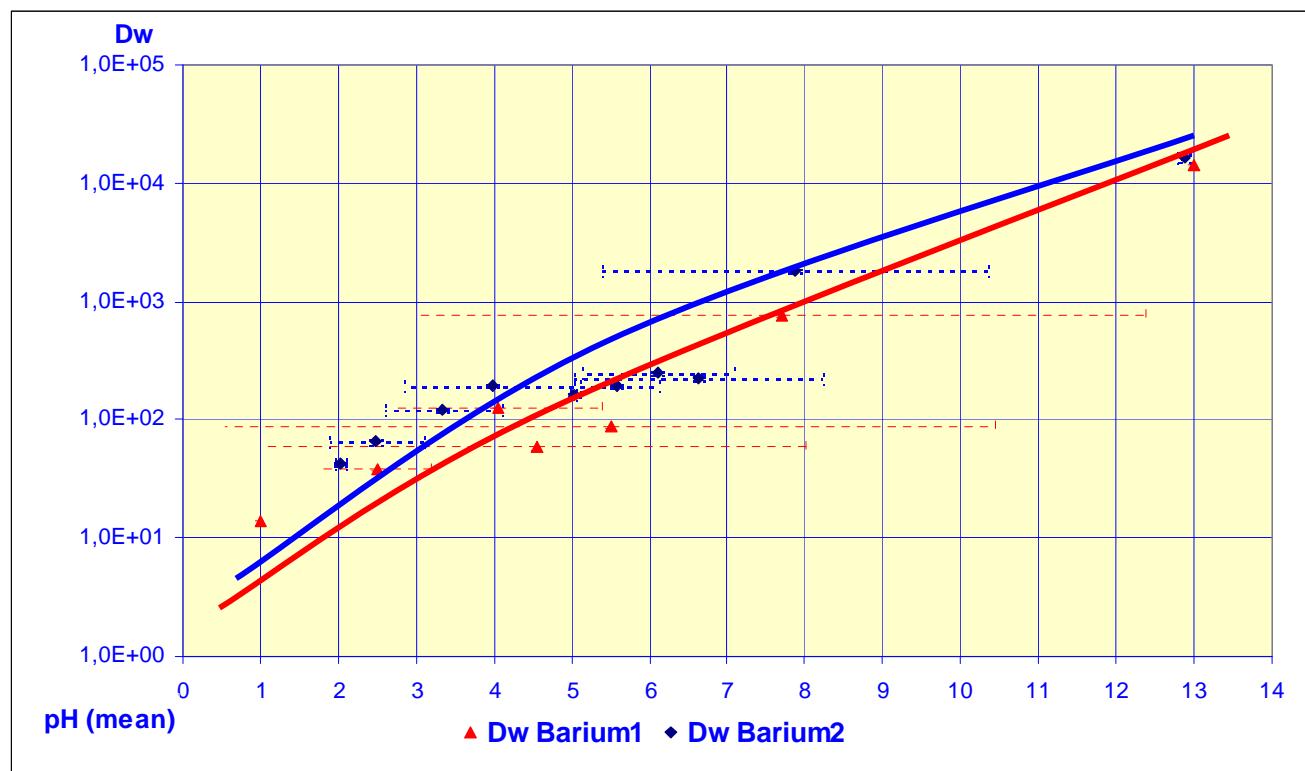
# D<sub>w</sub> - Ra-226



Moreno

- § High retention over the whole pH range
- § Radium adsorption greater than 99% for pH>3

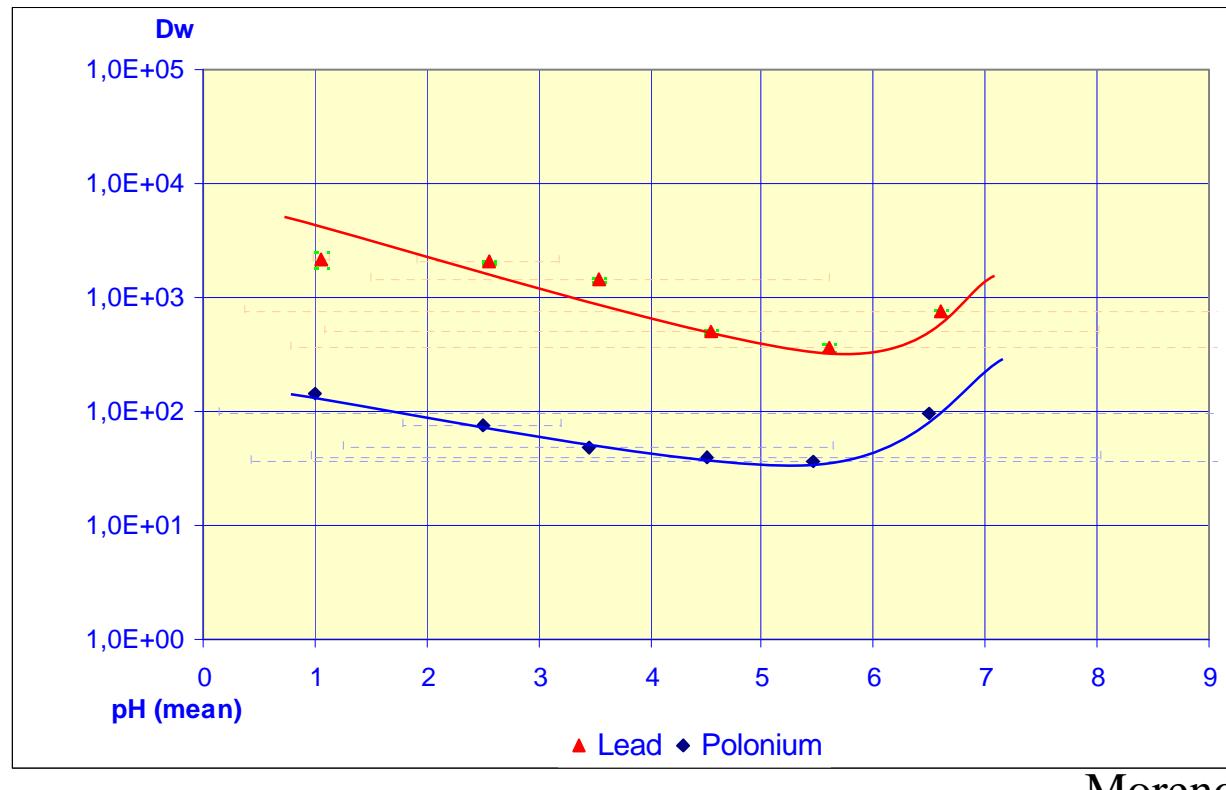
# D<sub>w</sub> - Ba-133



Moreno

- § Overall high retention above pH 4
- § Barium adsorption greater than 99% for pH>5

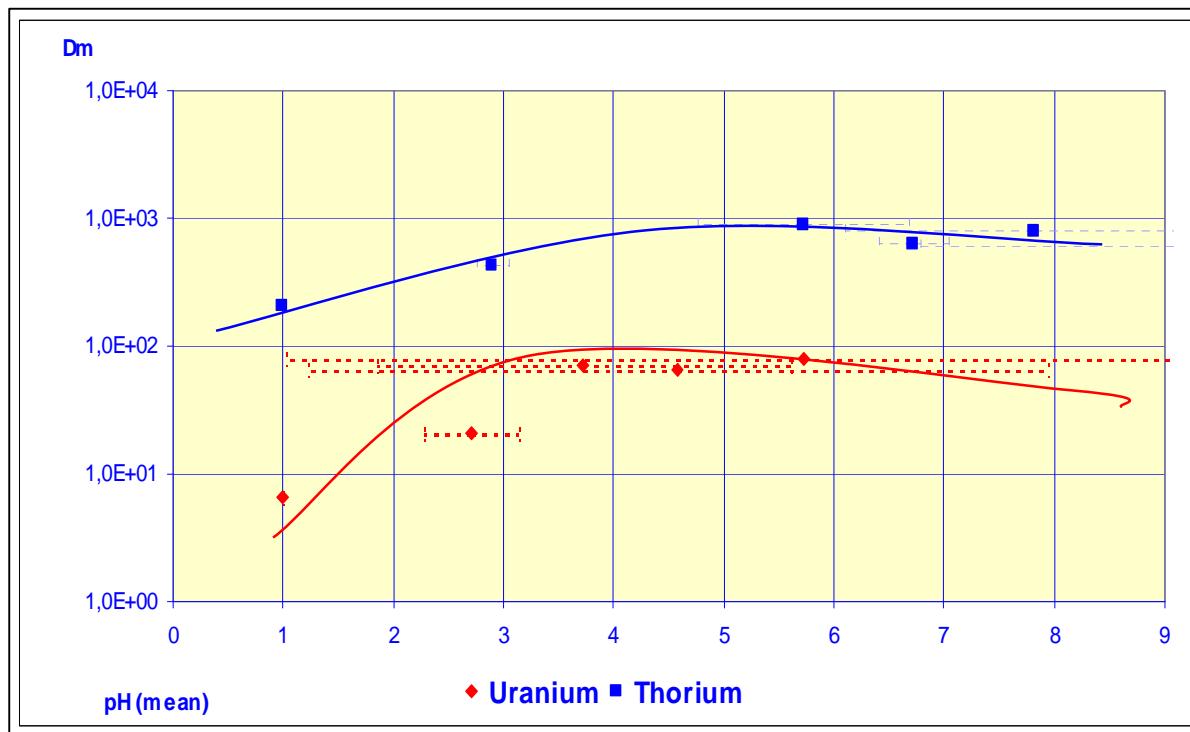
# D<sub>w</sub> - Pb-210/ Po-210



Moreno

- § Pb: High D<sub>w</sub> values ( $500 > D_w > 1000$ )
- § Po: Similar to the behaviour of Pb
- § D<sub>w</sub>(Po) overall lower (factor 10)

# $D_w$ - Th and U



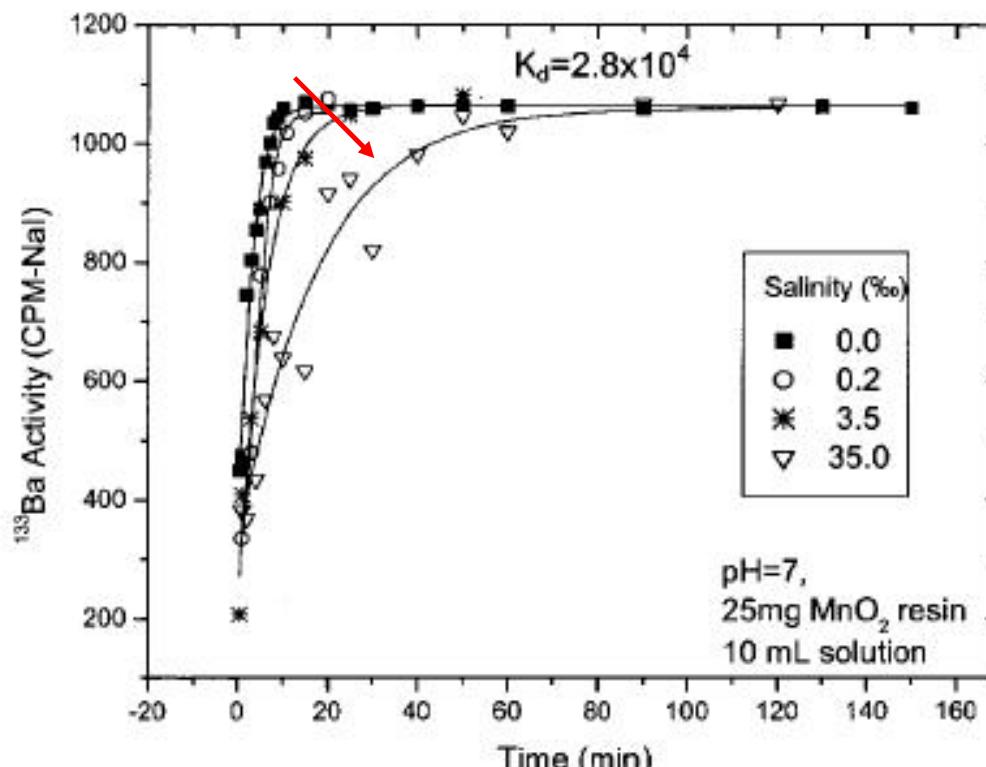
Moreno

- § U: rather good retention in near neutral pH range ( $D_w \bullet 100$ )
- § Th: overall very good retention  $300 < D_w < 1000$

## Kinetics

- § 10 mL water samples spiked with Ba-133 (0, 0.2, 3.5 and 35‰ salinities)
- § 25 mg MnO<sub>2</sub> resin
- § Magnetic stirrer
- § pH = 7.0
- § 1.0 to 90 minutes contact
- § Measurement of filtered MnO<sub>2</sub> in NaI well-type gamma counter

# Kinetics



Burnett et al.

§ Uptake kinetics depend on salinity

# Ba/Ra homology

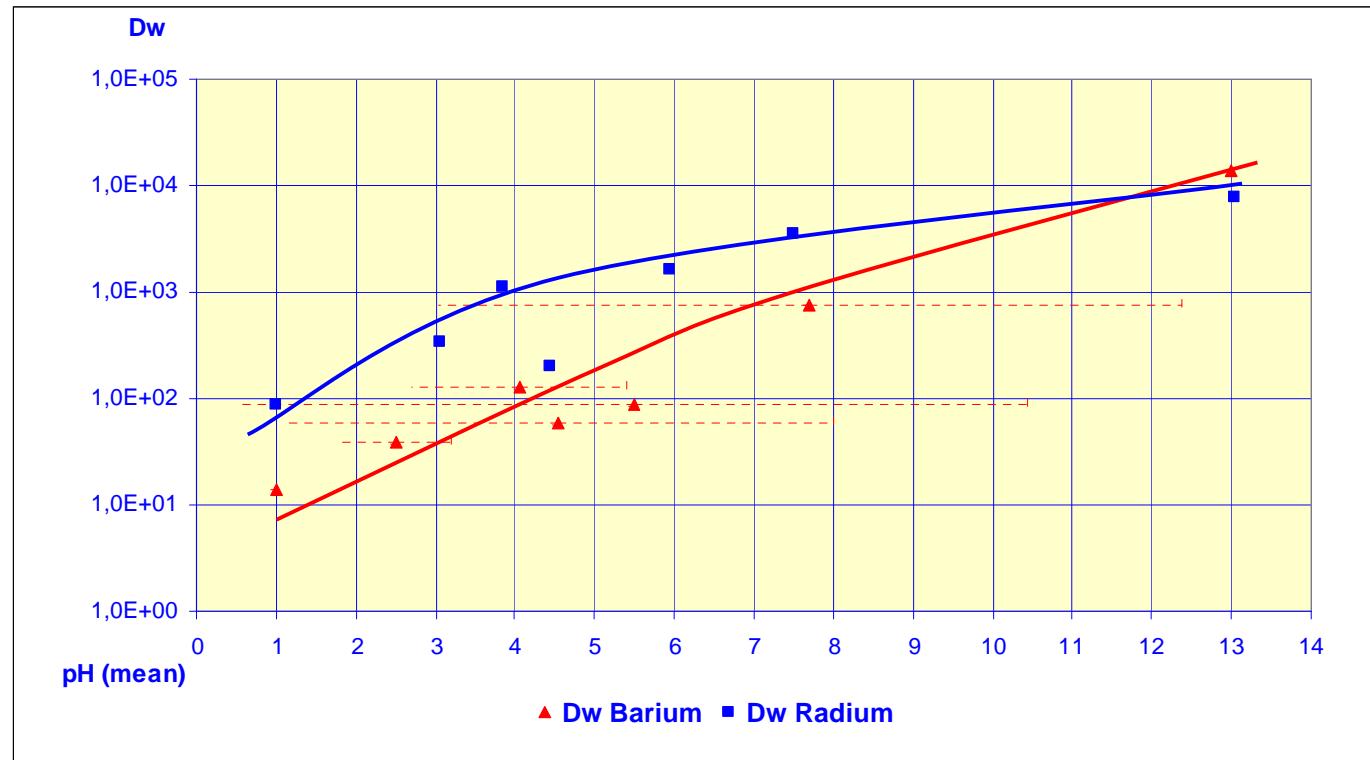
## § Batch experiments

- 100 mg MnO<sub>2</sub> / 5 mL solution

## § Column

- 1 g MnO<sub>2</sub> resin
- Geometry:  $\emptyset i = 0.9 \text{ cm}$ , H = 6,5 cm
- Water samples: de-ionised and synthetic sea water

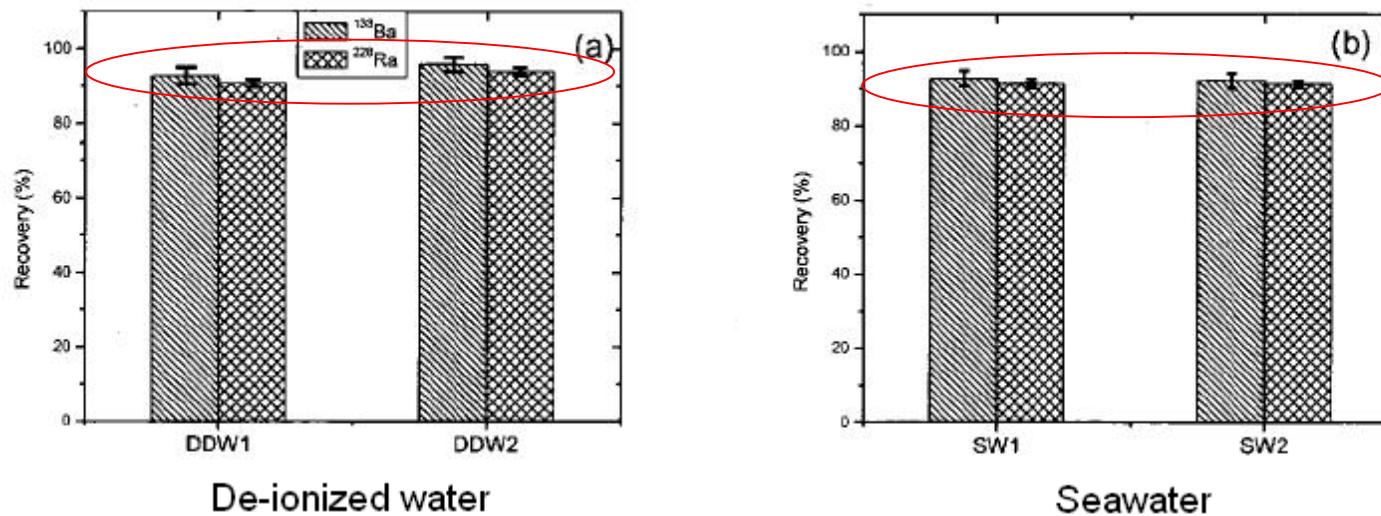
# Ba/Ra homology - batch



Moreno

- § High retention (>99%) above pH 5 for both
- § Good correspondence between Ra and Ba for pH values > 6-7
- § Significant deviation for low pH

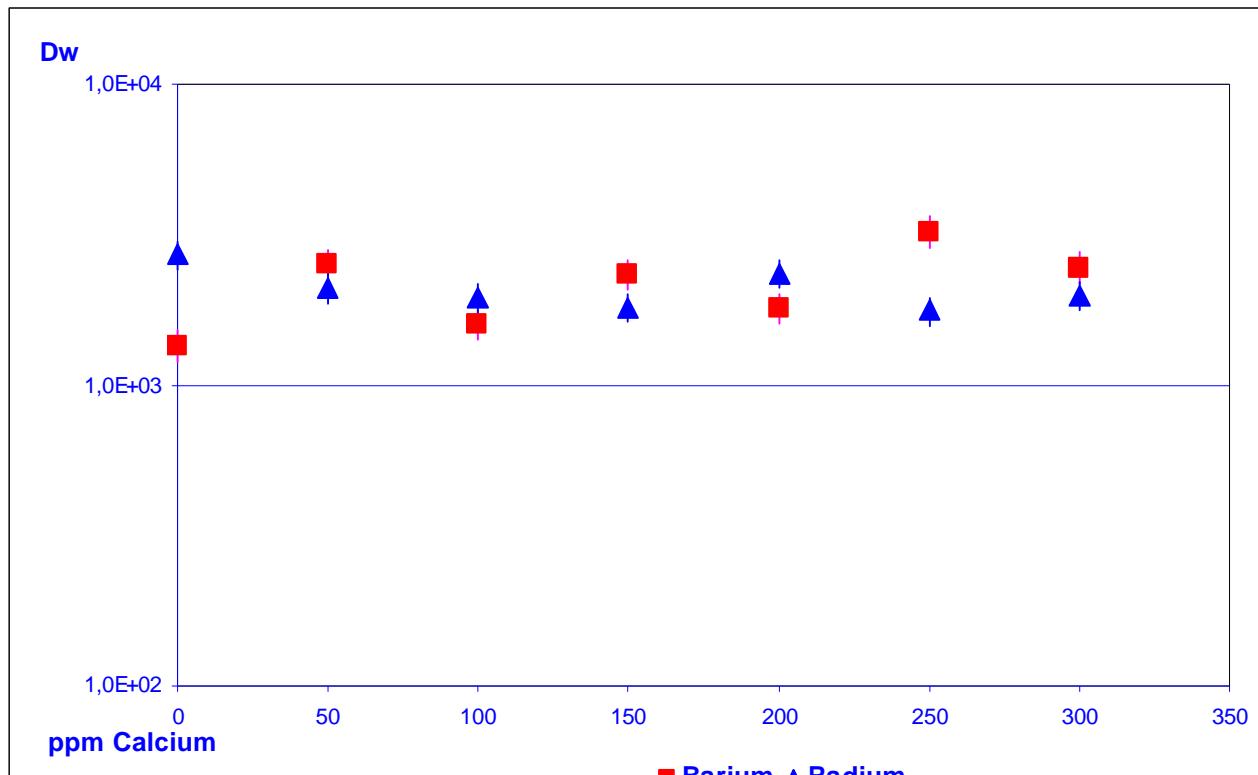
# Ba/Ra homology on columns



Burnett et al.

- § Good correspondence between Ba and Ra recoveries for de-ionized and Seawater
- § Recoveries > 90% for both matrices

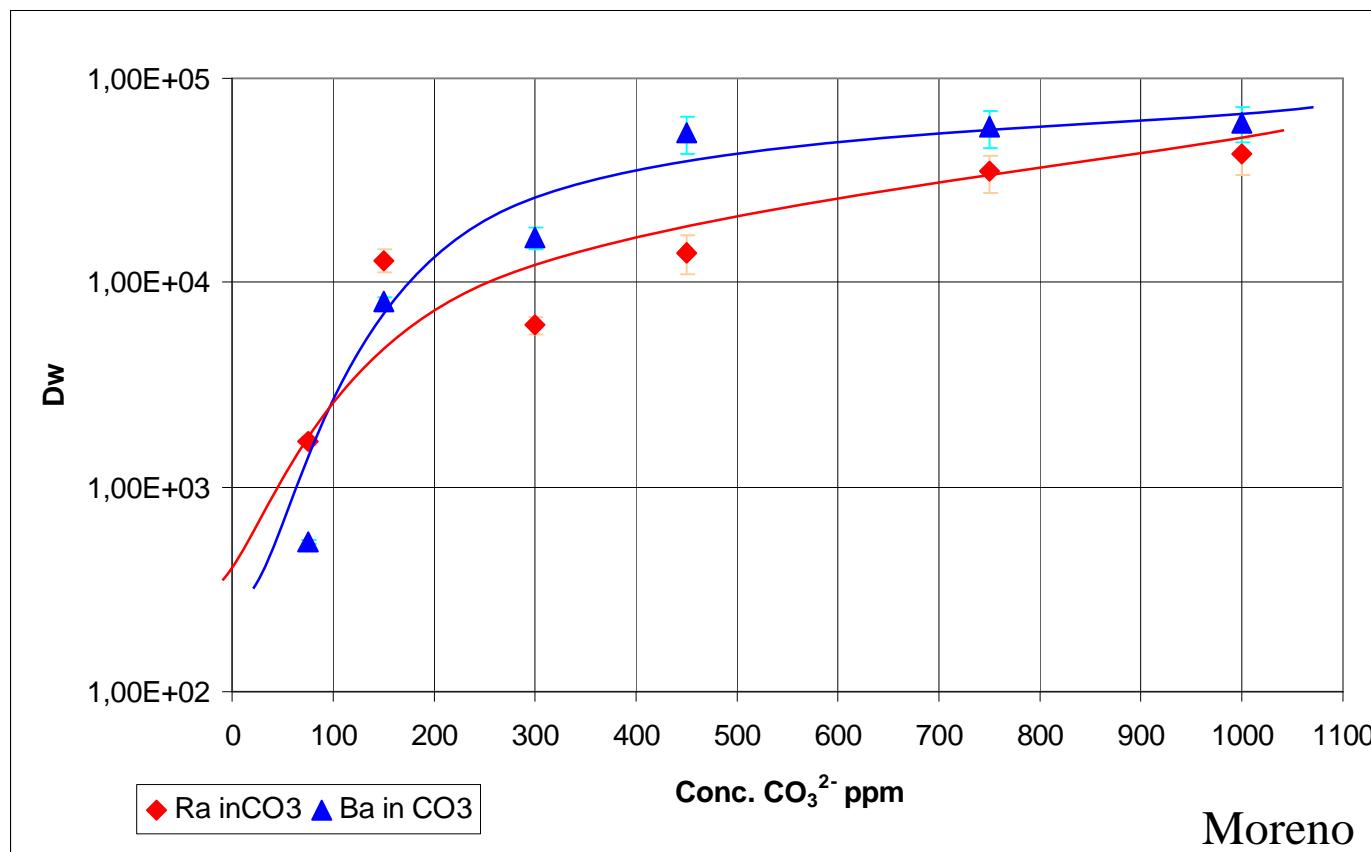
# Interferences - Ca



Moreno

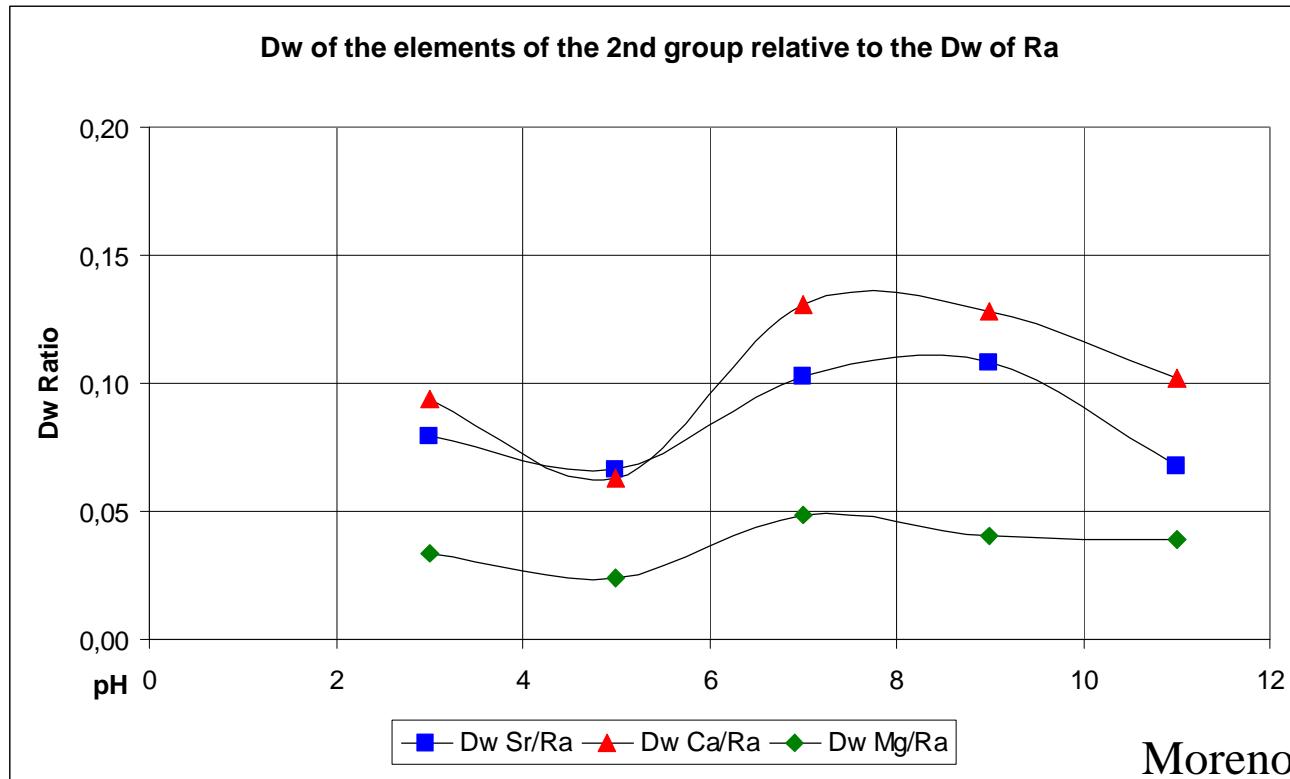
- § For [Ca] ≤ 300 ppm no interference on Ba and Ra retention (D<sub>w</sub>) observed

# Interferences - carbonate



- § Increase of  $D_w$  values with increasing carbonate concentration
- § Increase of carbonate concentration connected to increase of pH

# Selectivity over other earth-alkalines



- § Overall good selectivity for Ra over Sr, Ca and Mg
- § Sr and Ca show similar behaviour

# Capacities

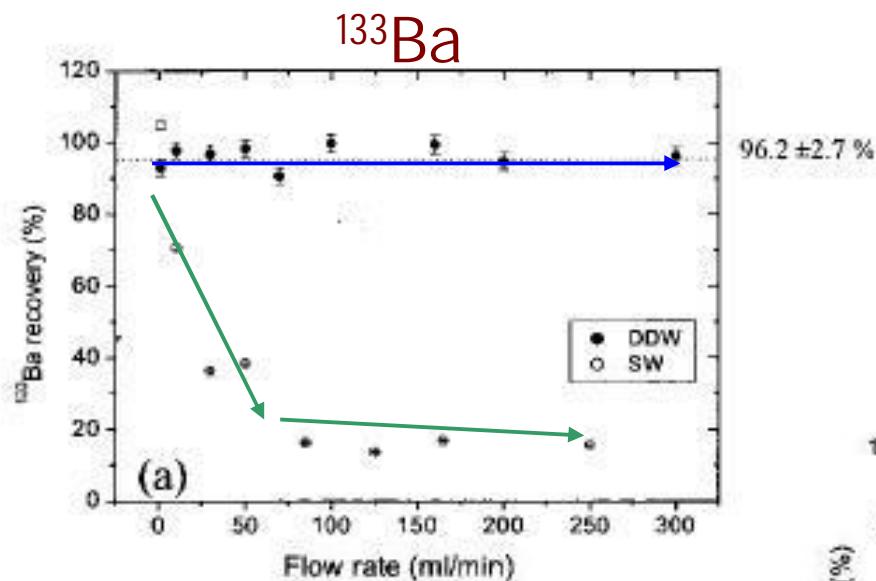
Element	Ba	Ba (300 ppm Ca)	Pb	Th
Capacity [mg/g resin]	~ 10	~ 10	~ 9	~ 1,0
pH value (mean)	5,6	4,5	6,5	5,2

- § Capacities in the order of 1 mg/g (Th) to 10 mg/g (Ba)
- § At indicated pH values
- § Capacities might differ when working at other pH values

## Flow rate

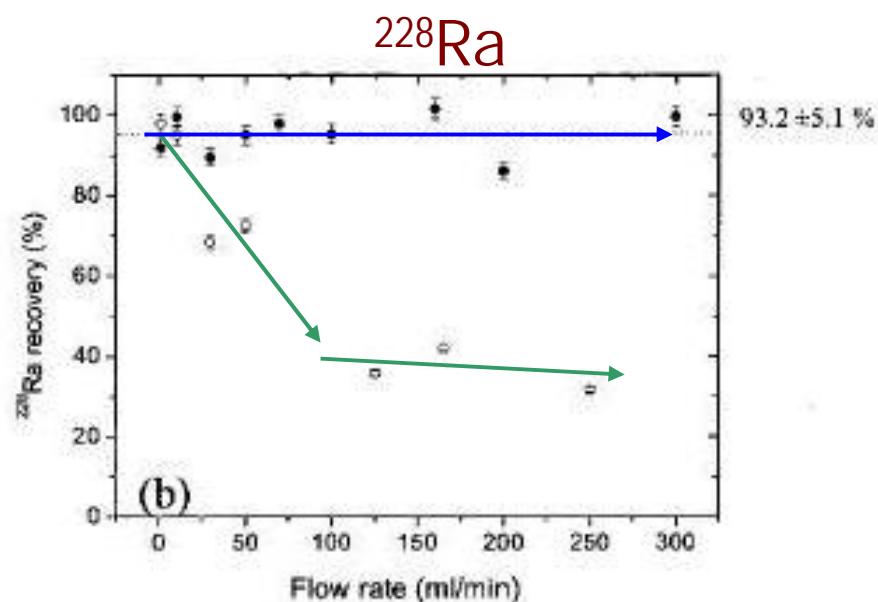
- § 1.0 L sample water spiked with  $^{133}\text{Ba}$  and  $^{228}\text{Ra}$
  - § 1.0 g  $\text{MnO}_2$  resin cartridge
  - § pH = 7.0
  - § Peristaltic pump, flow rate 0 -300 mL/min
  - § Samples: de-ionized and seawater
-

# Flow rates - results



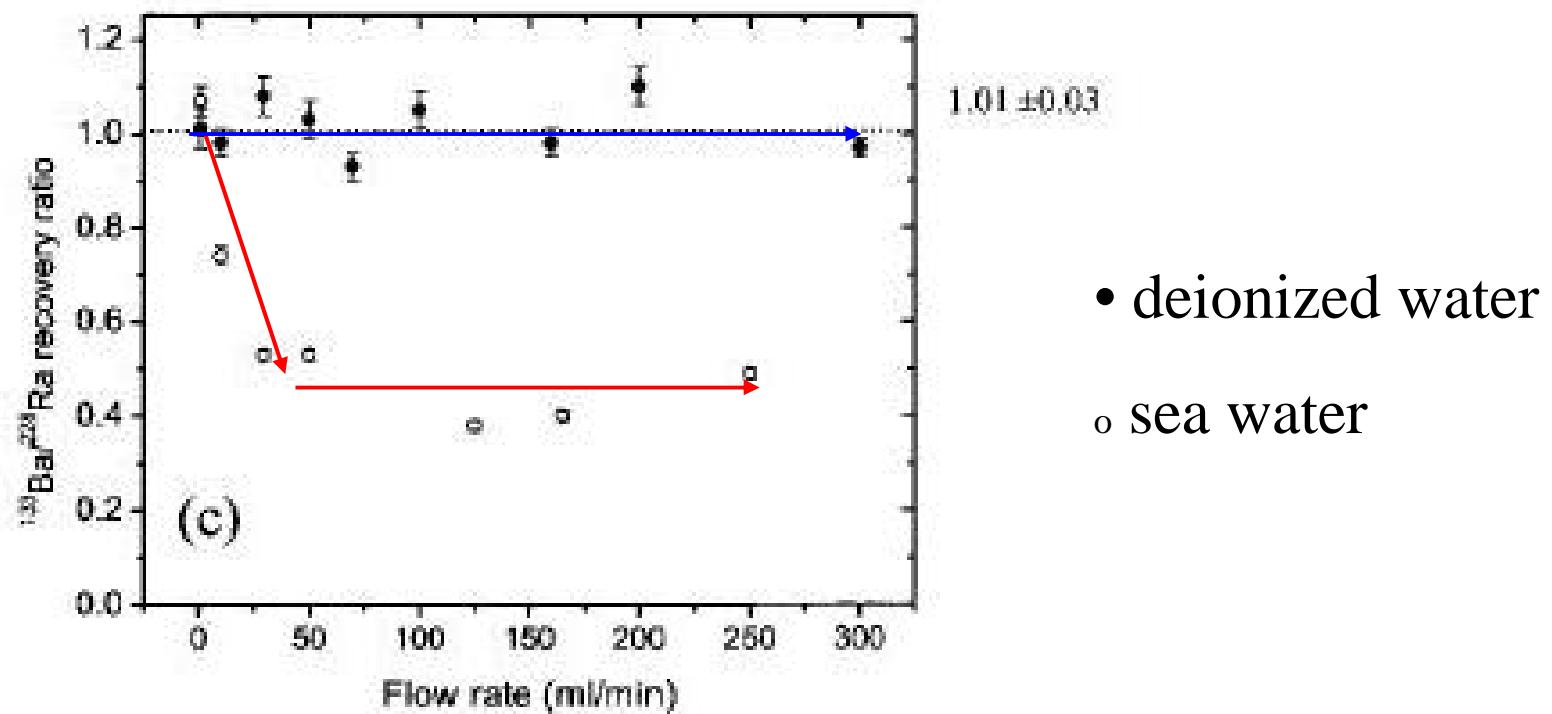
Burnett et al.

- deionized water
- sea water



∅ Problematic: Flow rates for sea water

# Flow rates – results: $^{133}\text{Ba}/^{228}\text{Ra}$



Burnett et al.

## Summary - Characterisation I -

- § Very high  $D_w$  ( $\sim 1000$ ) for Ra, Ba, Th and Pb
  - § Ba and Ra uptake kinetics depend on salinity
  - § Good Ba/Ra correspondence for pH >6
  - § No interference on Ba/Ra uptake from Ca and  $\text{CO}_3^{2-}$
-

## Summary - Characterisation II -

- § Good selectivity for Ra over Ca, Sr and Mg
- § Capacities between 1 mg/g (Th) and 10 mg/g (Ba)
- § Flow rates should be kept < 20 mL.min<sup>-1</sup> especially for high salt samples to assure:
  - High yields
  - Ba/Ra homology

# Applications

- § Ra-226/8 • updated two column draft procedure
    - MnO<sub>2</sub>/DGA
    - Ca rich water samples
    - Ac yield determination via gravimetry (CeF<sub>3</sub>)
  - § Preconcentration of Ba and Ra from 5L water samples for gamma spectrometry
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# New Ra-226/8 method

## § Why new method

- Increased interest in Ra isotopes (European DW directive)
- Current methods lengthy and time consuming or need long ingrowth times

## § Advantages of the new method

- High adsorption of Ra, Ra and other natural radionuclides onto MnO<sub>2</sub> resin from neutral pH solutions
- Rapid preconcentration from large water samples
- Selective / robust uptake and separation of Ac and other actinides on DGA Resin

## Ra-226/8 Procedure

### Ø Two Resin Method

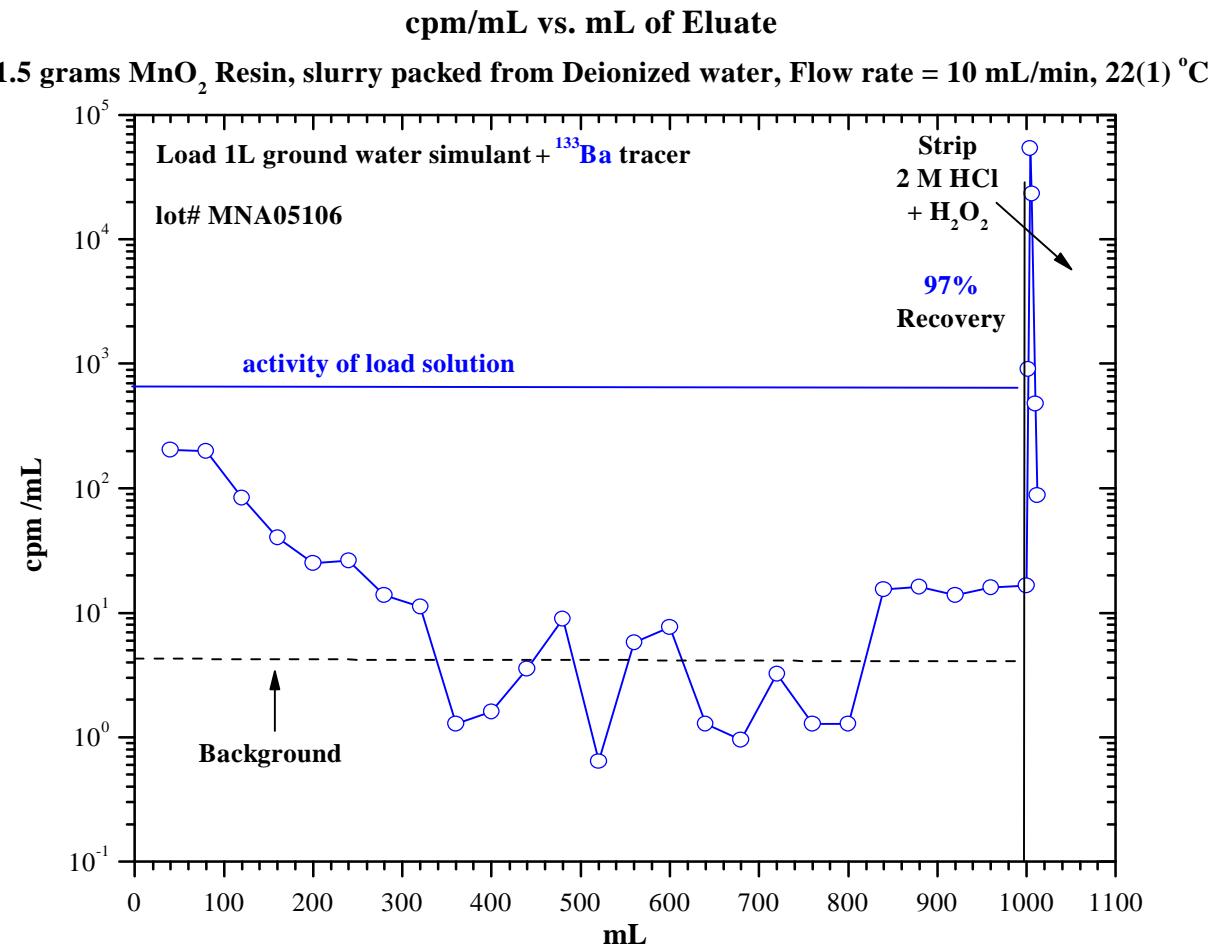
- MnO<sub>2</sub> Resin for pre-concentration, Ca removal
- DGA-Normal Resin for separation of Ac/Ce and Ra/Ba

- § Load sample (0,5 – 1L, pH 7) on 1g MnO<sub>2</sub>
- § Strip MnO<sub>2</sub> with 15 mL 5 M HCl-1.5% H<sub>2</sub>O<sub>2</sub>
- § Add Ce carrier (10 mg Ce)
- § Allow > 30 h for Ac-228 ingrowth
- § Load strip solution on DGA, Normal (2 mL)

# Ra-226/8 Procedure

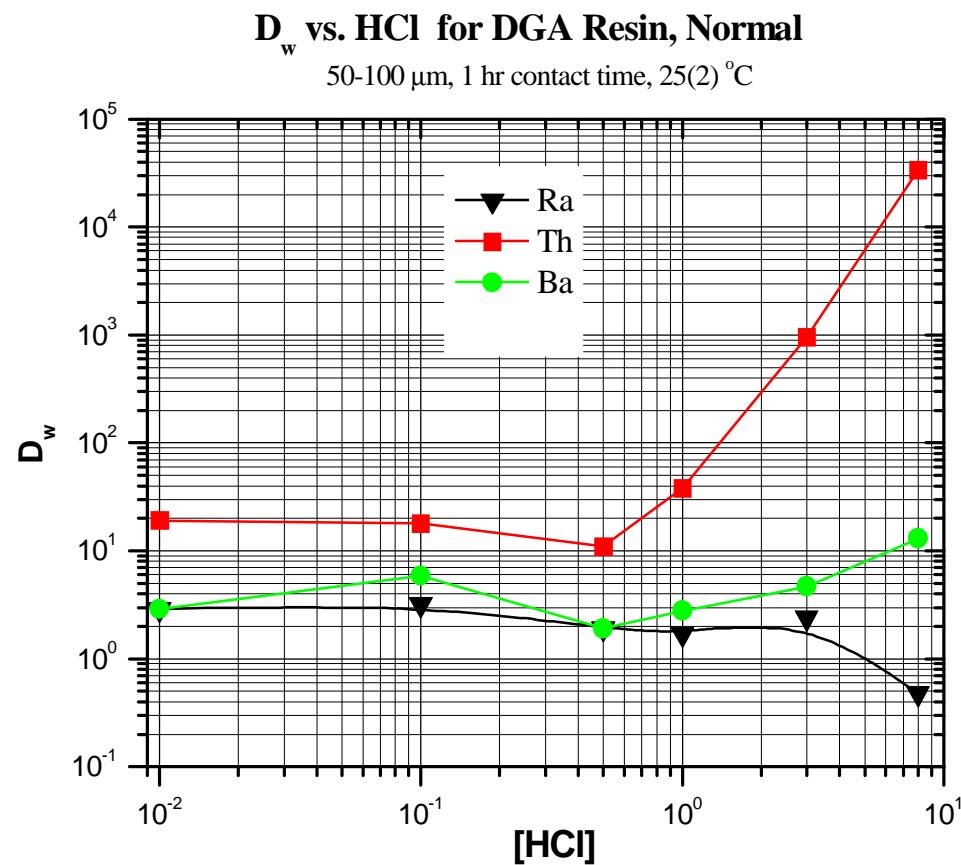
- § DGA: U, Th and Ac-228 retention, Ra and Ba pass
- § Collect load + 5 mL 5 M HCl rinse for Ra-226 and Ba-133
  - Microprecipitation with BaSO<sub>4</sub>
  - Yield by  $\gamma$ -spectrometry (Ba-133)
  - Ra-226 (and other  $\alpha$ -emitting Ra isotopes) by  $\alpha$ -spectrometry
- § Strip Ac-228 from DGA with 15 mL 2 M HCl
  - Microprecipitation with CeF<sub>3</sub>
  - Gas proportional counting
  - Yield determination Ac on DGA step via gravimetry
  - Yield of preconcentration step via Ba-133

# Ra/Ba preconcentration on MnO<sub>2</sub> resin



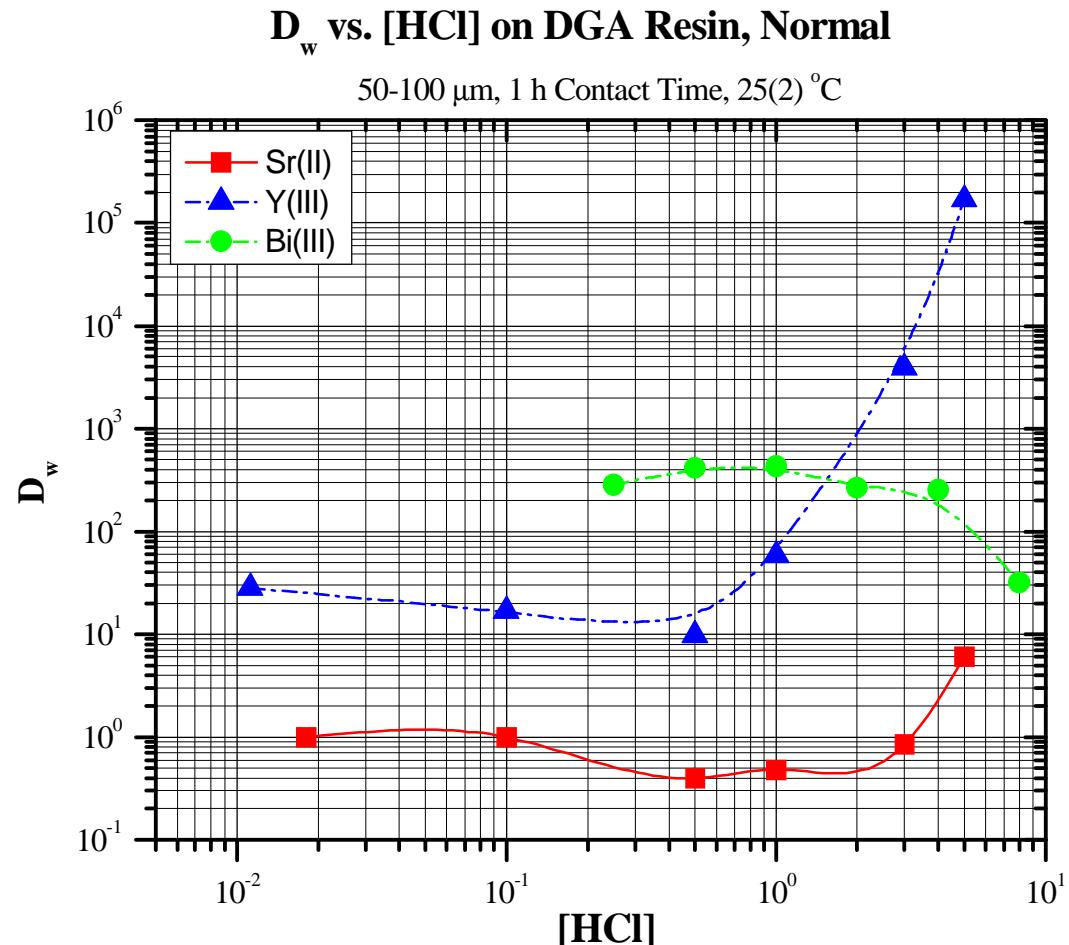
§ Elution profile Ba-133, 1L spiked ground water, 1.5 g MnO<sub>2</sub> resin

# Retention profiles - I



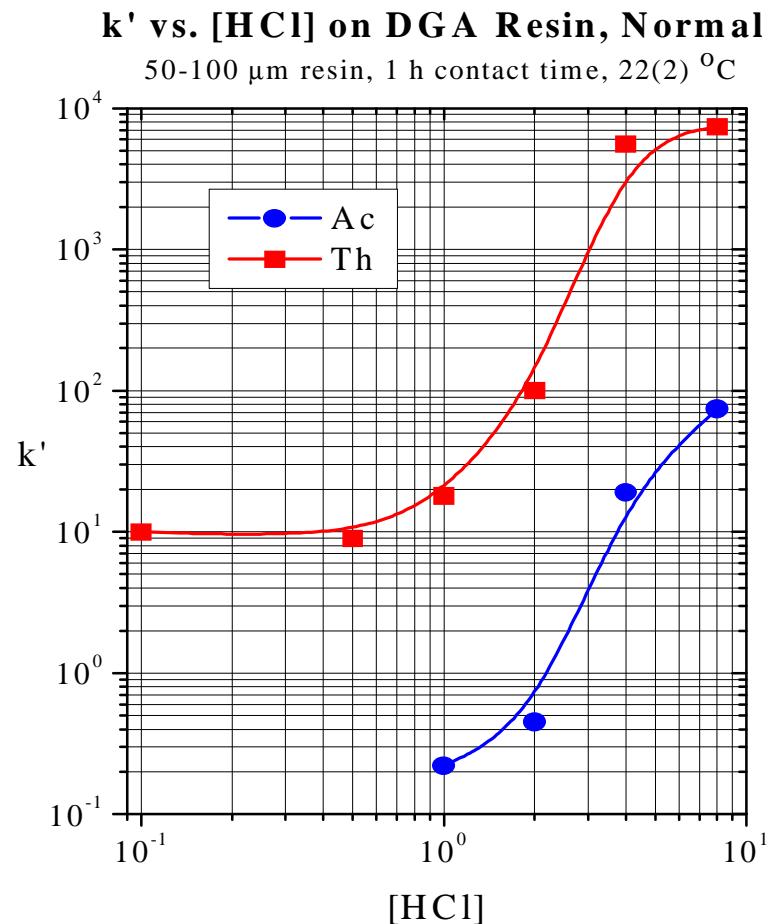
- § Retention of Ra, Th and Ba on DGA, normal in HCl
- § Strong Th uptake at high HCl concentration
- § Ba and Ra show low uptake

# Retention profiles - II



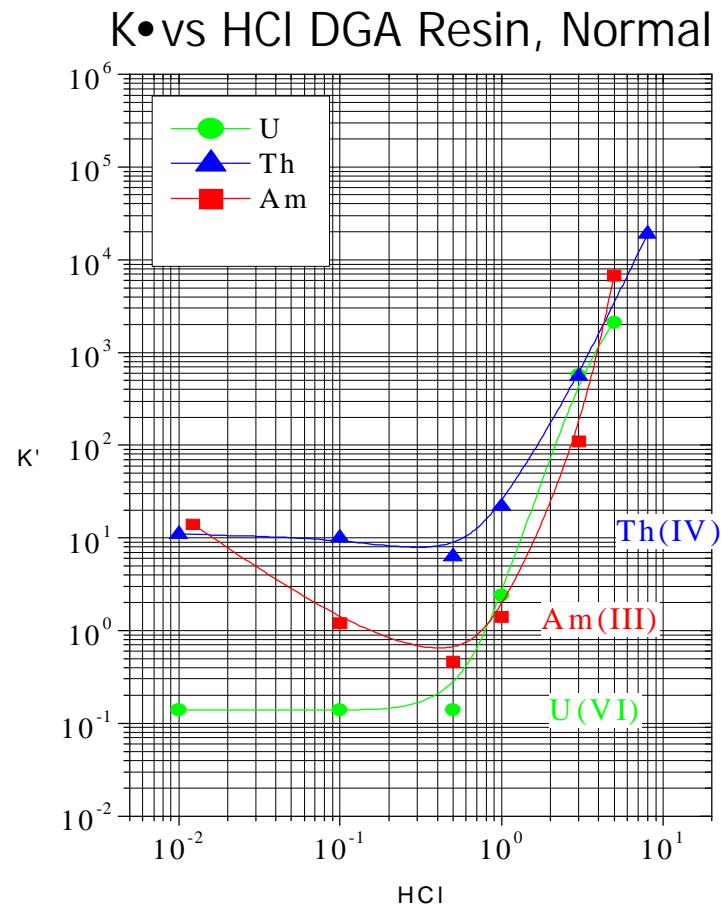
- § Retention of Sr, Y and Bi on DGA, normal in HCl
- § Y and Bi show high uptake at elevated HCl concentration
- § Potential interferents since chemistry similar to Ac

# Retention profiles - III



- § Retention of Ac and Th on DGA, normal in HCl
- § Ac: strong retention only for high HCl, significantly lower than Th
- § Th: overall good retention, especially for high HCl

# Retention profiles - IV



§ Retention of U, Th and Am  
on DGA, normal in HCl

§ Very similar for high HCl  
( $> 2$ M HCl)

# Measurement of Ba-133, Ac-228 and Ra

## § Ac-228

- Ac-228 sample preparation via cerium fluoride microprecipitation
- Counting on GPC

## § Ra and Ba-133

- Sample preparation via barium sulphate micro-precipitation
- Ba-133 counted by gamma spectrometry
- Ra isotopes counted by alpha spectrometry

# Decontamination Study

- § Selectivity of the DGA resin
- § Decontamination of commonly occurring alpha and beta emitters such as Pb/Bi, Th, Sr/Y were studied
- § 500 mL of tap water spiked with alpha and beta emitters
- § Samples processed using the complete method

Interference added	Deconfactor Ac-fraction
Th-228/U-232	>303
Pb-210/Bi-210	>1542
Sr-90/Y-90	>1208

∅ Overall very good decontamination from potentially interfering elements  
∅ Even for Y-90 and Bi-210 • MnO<sub>2</sub>?

# Method Performance Data

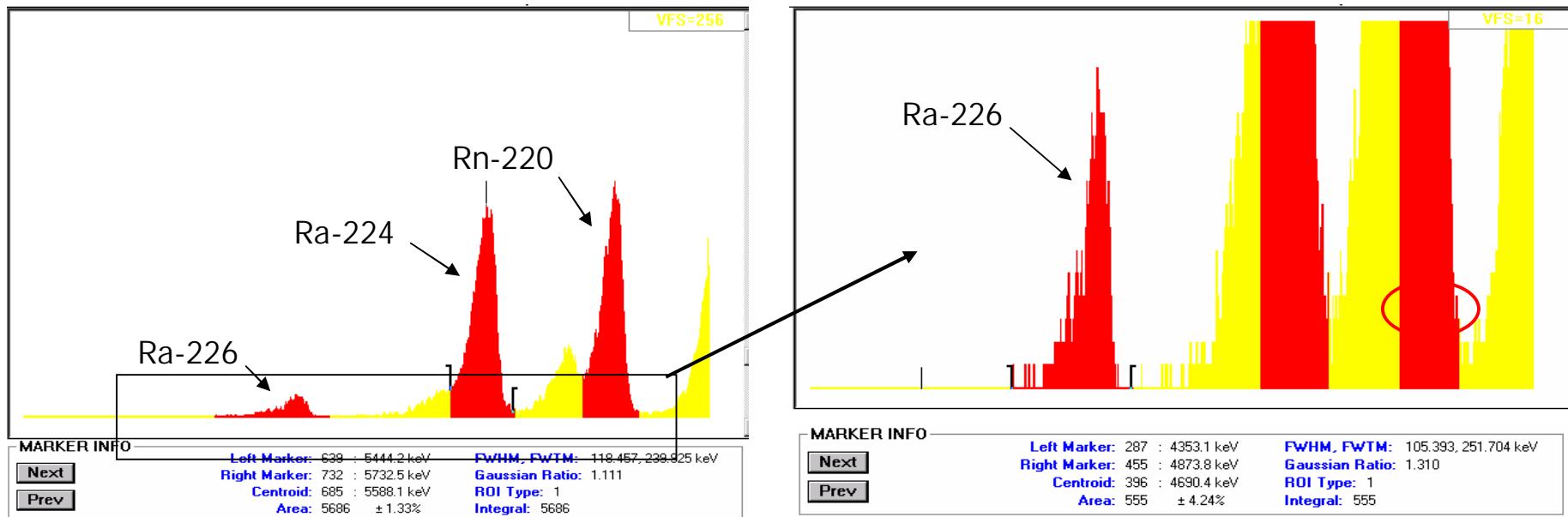
Replicate #	Yield Ba-133	Yield Ra-228	Yield ratio Ba/Ra	Yield bias / %	Yield Ra-226*
1	89	86	1,04	3,4	106
2	76	74	1,03	2,6	93
3	83	77	1,08	7,2	100
Average	83	79	1,05	4,8	99,7
SD	7	6	0,12	0,5	7
RSD / %	8,4	7,6	11,3	11,3	7,0

\*Yield corrected

# Method Performance Data

- § Good correspondence between Ba-133 and Ra-228 yields
  - Overall bias in the order of 5% • Ac-228 separation on DGA?
    - ∅ Bias not significant within given uncertainties
    - ∅ Correction via Ce yield for improved method
  - Ba-133 also internal standard for Ra-228/Ac-228
- § Corrected Ra-226 yield indicates absence of bias
- § Relative standard deviation (N=3) in the order of 7 – 8 %

# Alpha Spectra: Micro precipitation



1L, Spiked tap water (Ca • 500 ppm)

# Ba/Ra preconcentration for $\gamma$ -spectrometry

- § Special interest: Ra-228 determination
- § 5L water samples, pH 7
- § Ba-133 as internal standard
- § 3g MnO<sub>2</sub> Resin
- § Stirring over night at room temperature
- § Filtration, drying of the filter
- §  $\gamma$ -spectrometry after 48 h ingrowth period (Ac-228)

## Ba/Ra preconcentration for $\gamma$ -spectrometry

- § Ba-133 yield ~ 80%
- § Quicker, less hands-on time than evaporation / precipitation methods
- § Better reproducibility of source prep. for very Ca rich samples
- §  $DL_{Ra-228} (t = 4h, 20\% \text{ rel. Eff.}) = 60 \text{ mBq.L}^{-1}$  (*via* Ac-228, 911 keV)
- § No significant bias (Reference Material):

Determined activity A(Ra-228) / Bq.L-1	Uncertainty Uc (Ra-228) / Bq.L-1	Reference activity A(Ra-228) / Bq.L-1	Uncertainty Uc (Ra-228) / Bq.L-1	t-value
9,875	0,593	9,264	0,742	0,6

- § Extension to Pb-210 determination via  $\gamma$ -spectrometry?

## Summary - Applications -

First tests and results for applications:

- § Ra-226/8 via two column method
  - Excellent decontamination factors for potential interferents
  - No significant bias
- § Ra-228 determination via gamma spectrometry
  - Good yields
  - Application to larger volumes
  - Smaller amounts of MnO<sub>2</sub>